


Riga Technical University
Faculty of Material Science and Applied Chemistry



ABSTRACTS
of
Riga Technical University
54th International Scientific Conference

Section:
Material Science and Applied Chemistry
October 14-16, 2013, Riga, Latvia

Riga 2013

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Synthesis and Analysis of Novel Sugar Amino Acid-based Saccharopeptides

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¹Riga Technical University

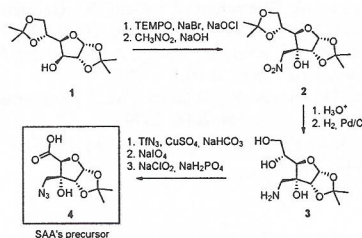
Keywords – glycoconjugates, oligosaccharide chemistry, sugar amino acids, foldamers, γ -peptides.

INTRODUCTION

Sugar amino acids (SAAs) represent a structurally diverse class of compounds that can be used in foldamer synthesis and as multifunctional monomeric building blocks for drug design. Oligomers of such monosaccharide hybrid molecules frequently adopt secondary structures in relatively short sequences. [1]

RESULTS AND DISCUSSION

We have developed a synthetic strategy towards a novel furanoid γ -SAA with *ribo* configuration. Commercially available diacetone- α -D-glucose **1** was used as a starting material in a multistep synthesis. The process started with TEMPO catalyzed oxidation affording corresponding ketone. [2] Further synthesis included a sequence of modifications at carbohydrate C(3) position. First, diastereomerically pure nitromethyl product **2** was prepared by the Henry reaction. After treatment with Brønsted acid and the reduction of the nitro group, deprotected aminomethylalcohol **3** was obtained. Azide functionality was introduced by diazotransfer reagent TfN₃. Finally, SAA precursor – γ -azidoacid **4** was prepared by two subsequent oxidation reactions, first with sodium periodate followed by Pinnick oxidation. Product **4** can be easily transformed into corresponding γ -amino acid by reduction of its azido group. Synthesis of **4** included six steps starting from diacetone- α -D-glucose **1** with an overall yield of 23% (Scheme 1).

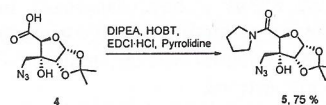


Scheme 1. Synthesis of SAA precursor **4**

Some derivatives of compound **4** were synthesised in order to obtain a fully protected monosaccharide unit

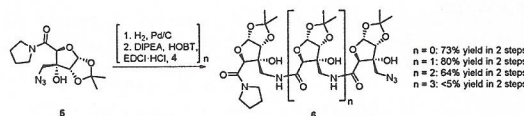
that could be used as a starting material in the synthesis of corresponding homooligopeptides. An appropriate orthogonal protection/deprotection strategy for liquid phase synthesis has been developed.

We have discovered that in our case the use of isopropyl ester as a protecting group for carboxyl termini is inappropriate since intramolecular cyclization occurs, once free amine is obtained. Hence, we focused on the linear homooligomer synthesis strategy. While the azido group was used as *N*-protection, the *C*-end was protected as a pyrrolidine amide (Scheme 2).



Scheme 2. Synthesis of SAA amide **5**

Compound **5** was further used for oligomer synthesis *via* standard solution phase peptide synthesis. Short-chain oligopeptides were obtained in adequate yields (Scheme 3).



Scheme 3. Oligopeptide synthesis

The presence of intramolecular hydrogen bond in the obtained SAA homooligopeptides was detected by ¹H-NMR-titration of their CDCl₃ and CD₃CN solutions with DMSO-*d*₆. It has been concluded that tetramer **6** (n=2) and pentamer (n=3) form at least one intramolecular hydrogen bond in CD₃CN solution.

We assume that further elongation of the obtained oligomers will lead to new artificial peptides that will form well-ordered secondary structures.

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New Approach

Irina Novosjolova¹, M

Keywords – purine nucleosides derivatives, 1,3-dipolar cyclo aromatic substitution, fluorescence

INTRODUCT

Modified purine derivatives anticancer and antiviral drugs significant applications also as of adenosine receptors [2].

Since 2002, many different nucleoside derivatives containing 1,2,3-triazole synthesised and investigated. Numerous literature reports deal with either (yl)-purine nucleosides [3]. In the process of synthesis of new adenine derivatives [4].

RESULTS AND DI

A straightforward way to the synthesis of (1-yl)-purine nucleosides developed from 2,6-diazidopurine with various terminal alkynes (for example 4-bromophenylacetylene, 4-tert-butyl-1-butyn-2-ol and others)

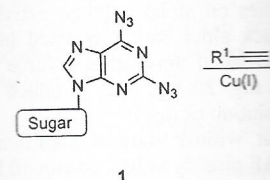


Figure 1. General synthesis of bis-

We have discovered that triazole nucleophilic substitution with nucleophiles. The substitution proven by X-ray analysis (Figure 1).

Recently, we have discovered a new nucleophile we obtain C(2) substituted following "click" reaction giving